

## NEW NATURAL DITERPENE ACIDS FROM *JUNIPERUS COMMUNIS*\*

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**Key Word Index**—*Juniperus communis*; Cupressaceae; diterpenoids; new labdane and pimarane derivatives.

**Abstract**—Three new diterpene acids have been isolated from the leaves of *Juniperus communis* and their structures, elucidated by spectroscopic methods, were identified as 7-oxo-13-*epi*-pimara-8,15-dien-18-oic acid, 7 $\alpha$ -hydroxysandaracopimaric acid and (14*S*)-14,15-dihydroxylabda-8(17),13(16)-dien-19-oic acid. Biflavonyls, fatty acids and diterpenoids with known structures were also isolated.

### INTRODUCTION

In previous papers of this series we have reported results on the composition of both volatile oil and non-volatile extracts from berries of common juniper [1–3] and other *Juniperus* spp. [4, 5]. In another report we examined the composition of leaves of *Juniperus oxycedrus* [6]. In this paper we report the isolation and structural determination of components of an ethereal extract from *Juniperus communis* leaves.

### RESULTS AND DISCUSSION

The ether extract of air-dried leaves was dewaxed and fractionated with aqueous NaHCO<sub>3</sub>, Na<sub>2</sub>CO<sub>3</sub> and NaOH solutions and the acidic fractions were worked up separately. From the NaOH fraction *trans*-communic acid (**1**), isopimaric acid (**2**) and a mixture of 12-hydroxylauric acid and 16-hydroxypalmitic acid were isolated. The Na<sub>2</sub>CO<sub>3</sub> contained isopimaric acid (**2**), sandaracopimaric acid (**3**), imbricatolic acid (**4**), isocupressic acid (**5**), and 13,14-epoxyimbricatolic acid (**6**), which had also been isolated from *Juniperus thurifera* [7] and through periodic acid oxidation gave the dinor-compound **7**, another component of *J. thurifera*.

The NaHCO<sub>3</sub> fraction contained 7 $\alpha$ -hydroxydehydroabiatic acid (**8**), two biflavonyls, hinokiflavone and cupressuflavone [6] and the three new diterpenoids, 7-oxo-13-*epi*-pimaric-8,15-dien-18-oic acid, (14*S*)-14,15-dihydroxylabda-8(17),13(16)-dien-19-oic acid and 7 $\alpha$ -hydroxysandaracopimaric acid which were isolated as methyl esters (**9**, **15** and **16**, respectively). The structures of all known compounds were established by physical and spectroscopical data and confirmed by comparison with authentic samples.

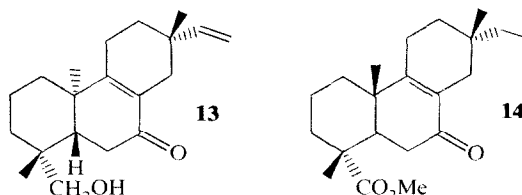
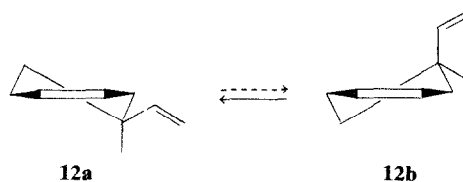
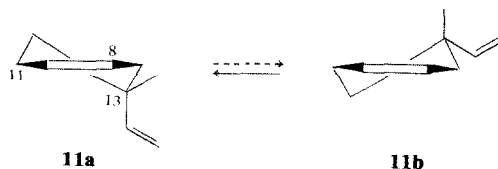
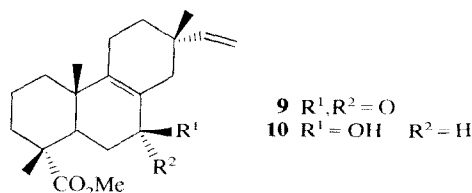
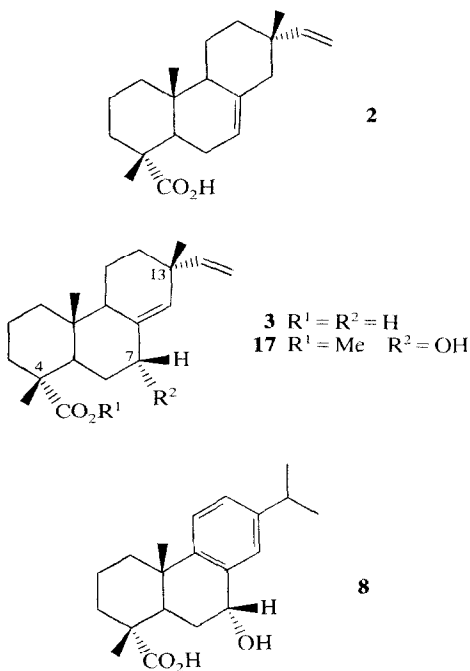
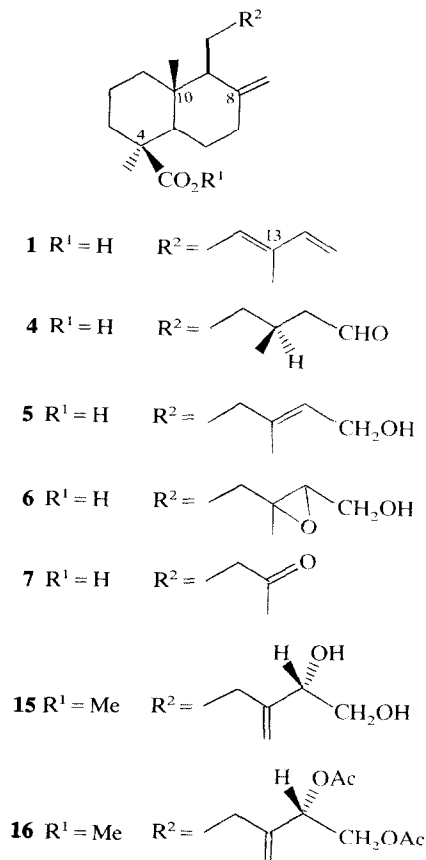
#### Methyl 7-oxo-13-*epi*-pimara-8,15-dien-18-oate (**9**)

The IR spectrum of **9** showed bands at 1725 and 1240 cm<sup>-1</sup> which confirmed the presence of an equatorial esterified carboxyl group [8], at 1660 and 1610 cm<sup>-1</sup> ( $\alpha,\beta$ -unsaturated carbonyl) and at 3070, 990 and 905 cm<sup>-1</sup> (vinyl substituent). The UV spectrum has an absorption maximum at 244 nm (log  $\epsilon$  4.05) and suggested a trisubstituted conjugated double bond. The <sup>1</sup>H NMR spectrum included signals of three quaternary methyl groups ( $\delta$  1.00, 1.12 and 1.27) and those of a typical ABX system of a vinyl group (4.69–5.00, 2H, *m* and 5.50–5.92, 1H *sextet*) but no other resonance of olefinic protons was present. All these data are consistent with a  $\Delta^{8,15}$ -pimaradiene skeleton with a C<sub>18</sub>-equatorial carboxyl group and a keto group conjugated with a  $\Delta^8$ -unsaturation. The carbonyl must be placed at C-7 because the NaBH<sub>4</sub> reduction of **9** gave a hydroxy compound, **10**, showing a triplet at  $\delta$  4.10 for the geminal proton to the OH group (thus excluding the C-14 position), and because the carbonyl or  $\beta$ -hydroxyl groups at C-11 should induce on the C-10 methyl group, larger shifts than those observed (1.12 and 1.06 for **9** and **10**) [9] in both substances.

As shown by <sup>13</sup>C NMR spectroscopy [10], the conformation of ring C in  $\Delta^8$ -pimarenes must be **11a** or **12a**, regardless of the configuration at C-13 and the free energy calculations based on data for isolated cyclohexanes. Furthermore, compounds **9** and **10** do show in the <sup>1</sup>H NMR spectrum a sextet for the X part of the vinyl ABX system. This unusual multiplicity has been observed when the vinyl group is axial in a cyclohexene but not when it is equatorial [11, 12]. Hence, the vinyl group must be axial and the relative configuration as shown in **11a**.

The CD of **9** ( $\Delta\epsilon_{339} + 0.84$ ,  $\Delta\epsilon_{249} - 0.77$ ,  $\Delta\epsilon_{213} + 3.52$ ) showed maxima opposed in sign to those of **13** ( $\Delta\epsilon_{339} - 0.5$ ,  $\Delta\epsilon_{246} + 1.97$ ,  $\Delta\epsilon_{210} - 2.42$ ), a compound recently isolated by Bohlmann *et al.* [13]. These data and the established rules for the CD of enones [14, 15] led us to conclude that the ketoester **9** belongs to the

\* Preceding paper in this series: Pascual Teresa, J. de *et al.* (1978) *An. Quím.* **74**, 1093.



be rationalized if some coupling between the axial vinyl and the enone  $\pi$ -electrons is assumed [16].

Lastly, the structure of **9** was confirmed by partial hydrogenation to yield **14**, a substance with identical IR and comparable (different solvent)  $^1\text{H}$  NMR spectra to those of the compound previously synthesized by Herz and Hall [17].

#### Methyl-(14 *S*)-14, 15-dihydroxylabda-(17), 13(16)-dien-19-oate (**15**)

The molecular formula of **15**,  $\text{C}_{21}\text{H}_{34}\text{O}_4$ , was deduced from the MS data of its diacetate. The IR spectrum showed intense bands of hydroxyl ( $3400\text{ cm}^{-1}$ ) vinylidene ( $3070$ ,  $1640$ ,  $890\text{ cm}^{-1}$ ) and axial carboxylic ester groups ( $1725$ ,  $1230$ ,  $1190$ ,  $1160\text{ cm}^{-1}$ ) [8]. The  $^1\text{H}$  NMR spectrum showed signals of three methyl singlets ( $\delta$  5.1, 1.20 and 3.61), a hindered multiplet of two methylene protons of a primary alcohol (3.65), another multiplet (4.14) of a proton geminal with a secondary allylic hydroxyl group, two broad singlets (4.50 and 4.86) of an exocyclic methylene and two vinylidene protons at 4.95 and 5.12. The  $^1\text{H}$  NMR spectrum of the diacetate **16** showed an ABX system (3.72–4.33 AB part and 5.14–5.36 X part) of the geminal protons to both acetoxy groups which, consequently, should be vicinal. These data suggested that **15** had a labdane structure. As a definitive proof of the structure, a substance with identical spectral properties was obtained from methyl isocupressate by oxidation with singlet oxygen. The CD of diol **15** measured in  $\text{CCl}_4$  containing  $\text{Pr}(\text{fod})_3$  showed a negative dichroic absorption ( $\Delta\epsilon_{317} - 1.2$ ) which led us to assign the configuration *S* for C-14 [18].

normal  $5\alpha$ -pimarane series, like all the other pimaranes isolated from *J. communis*, and so the absolute configuration at C-13 must be *S*.

The CD curve of **9** was also in agreement with the axial conformation of the vinyl group. The smaller dichroic absorption of **9** relative to **13** at 246 nm could

The same sign of the CE is shown by a halimic acid derivative with the same side-chain ( $\Delta^{13(16)}$ , (14S)-hydroxydihydrohalimic acid [19]), whose stereochemistry was also established by Horeau's method.

#### Methyl 7 $\alpha$ -hydroxysandaracopimarate (17)

The MS of **17** showed the  $M^+$  at  $m/e$  332, in agreement with the molecular formula  $C_{21}H_{32}O_3$ . The IR spectrum showed bands of a hydroxyl group, vinyl unsaturation and an equatorial carboxyl group. The  $^1H$  NMR spectrum had four methyl singlets at  $\delta$  0.84, 1.08, 1.22 and 3.70 and the absorption pattern of vinylic protons was identical with that of sandaracopimaric acid (**3**). The multiplet at 4.16 ( $W_{1/2} = 9$  Hz) of the geminal proton to the secondary axial OH group and the observed deshielding of 0.25 ppm for the proton at C-14 suggested the localization of the hydroxy group at C-7. These properties are identical to those of a photooxidation product of methyl isopimarate (**2** methyl ester) described by Fourrey *et al.* [20]. Another substance related to **17**, 7 $\alpha$ -hydroxy-(-)-pimara-8(14),15-dien-19-oic acid, was isolated by Yanagisawa *et al.* [21] from *Aralia cordata*.

#### EXPERIMENTAL

Optical rotations were measured in  $CHCl_3$  soln, IR spectra in film,  $^1H$  NMR (60 MHz) generally in  $CDCl_3$  with TMS as int. standard; chemical shifts are in ppm. MS at 70 eV. Si gel G and Si gel 60 (Merck) were used in PLC and CC separations.

**Extraction and isolation.** *Juniperus communis* (female gametophyte) was collected in Soria (Spain), September 1974. Dried leaves (1.5 kg) were extracted with refluxing  $Et_2O$  for 48 hr. Dewaxing with MeOH gave waxes (44.8 g) and the remaining product in  $Et_2O$  soln was extracted with aq. satd  $NaHCO_3$  and 4% NaOH solns to yield, respectively, 17.8, 21.5 and 4.4 g, leaving 24.3 g of neutral fraction. The NaOH fraction after CC on Si gel (150 g) yielded *trans*-communic acid (**1**) (0.62 g eluted with  $C_6H_6$ ), isopimaric acid (**2**) (0.95 g,  $C_6H_6-Et_2O$ , 9:1), fatty acids (1.3 g,  $C_6H_6-Et_2O$ , 9:1 and 8:2), 12-hydroxylauric acid and 16-hydroxypalmitic acid (0.8 g,  $C_6H_6-Et_2O$ , 1:1). All compounds had properties as described previously [2]. The  $Na_2CO_3$  fraction by CC on Si gel (500 g) yielded **1+2** (5.7 g  $C_6H_6$ ), **2**+sandaracopimaric acid (**3**) (5.15 g,  $C_6H_6-Et_2O$ , 9:1), imbricatic acid (**4**) (2.24 g,  $C_6H_6-Et_2O$ , 9:1), isocupressic acid (**5**) (5.9 g,  $C_6H_6-Et_2O$ , 7:3) and, after esterification and CC, methyl 13,14-epoxyinbricatolate (**6**) (21 mg  $C_6H_6-Et_2O$ , 9:1). Properties as described previously [3, 7]. The  $NaHCO_3$  fraction by CC on Si gel (550 g) yielded **2**(2.8 g), fatty acids (2.6 g), **5**(1.61 g), methyl 7-oxo-13-epi-pimara-8,15-dien-18-oate (**9**) (123 mg after esterification and new CC,  $C_6H_6-Et_2O$ , 9:1), methyl-(14S)-14,15-dihydroxylabda-8(17),13(16)-dien-19-oate (**15**) (125 mg after esterification and CC,  $C_6H_6-Et_2O$ , 8:2), hinokiflavone (23 mg by PLC on Si gel,  $MeC_6H_5-HOAc-Py$ , 4:1:1), cupressuflavone (152 mg,  $C_6H_6-EtOAc$ , 3:7), methyl-7 $\alpha$ -hydroxydehydroabietate (8 Me ester) (40 mg after esterification and PLC purification  $C_6H_6-Et_2O$ , 9:1) and methyl-7 $\alpha$ -hydroxysandaracopimarate (**17**) (77 mg, purified as **8**).

**Methyl 7-oxo-13-epi-pimara-8,15-dien-18-oate (9).** Colourless oil.  $C_{21}H_{30}O_3$ ;  $[\alpha]_D +77.6^\circ$  (c 0.76); IR  $\nu_{max}$   $cm^{-1}$ : 3070, 1725, 1660, 1610, 1240, 990, 905; UV  $\lambda_{max}$  nm (log  $\epsilon$ ): 244 (4.05);  $^1H$  NMR:  $\delta$  1.00 (3H, s, Me-13), 1.12 (3H, s, Me-10), 1.27 (3H, s, Me-4), 3.61 (3H, s, OMe),

4.69–5.00 (2H, m,  $=CH_2$ ), 5.50–5.92 (1H, m,  $-CH=$ ). CD (*n*-hexane):  $\Delta\epsilon_{330} +0.84$ ,  $\Delta\epsilon_{249} -0.77$  and  $\Delta\epsilon_{213} +3.52$ .

**Methyl-7 $\alpha$ -hydroxy-13-epi-pimara-8,15-dien-18-oate (10).** To a MeOH soln of **9** (46 mg),  $NaBH_4$  (10 mg) was added and refluxed for 10 min. Recovery of the product as usual gave **10** (41 mg).  $C_{21}H_{32}O_3$ , oil,  $[\alpha]_D +42.9^\circ$  (c 0.67); IR  $\nu_{max}$   $cm^{-1}$ : 3400, 3070, 1725, 1640, 1240, 990, 905;  $^1H$  NMR:  $\delta$  1.00 (3H, s, Me-13), 1.06 (3H, s, Me-10), 1.22 (3H, s, Me-4), 3.62 (3H, s, OMe), 4.10 (1H, t,  $J = 8$  Hz, H-7), 4.60–4.98 (2H, m,  $=CH_2$ ) and 5.46–5.92 (1H, m,  $-CH=$ ).

**Methyl 7-oxo-13-epi-pimara-8-en-18-oate (14).** **9** (65 mg) was hydrogenated in  $C_6H_6$  soln at atm. pres. in the presence of  $PtO_2$  (28 mg) for 5 hr. After PLC on Si gel, **14** (35 mg) was obtained.  $[\alpha]_D +17.4^\circ$  (c 1.09); IR  $\nu_{max}$   $cm^{-1}$ : 1720, 1660, 1610, 1240;  $^1H$  NMR ( $CCl_4$ ):  $\delta$  0.82 (3H, s, Me-13), 0.83 (3H, t,  $J = 7$  Hz, H-16), 1.11 (3H, s, Me-10), 1.22 (3H, s, Me-4), 3.64 (3H, s, OMe). MS  $m/e$  (rel. int.): 332 ( $M^+$ , 7), 303 (8), 289 (6), 273 (4), 91 (100). CD (*n*-hexane)  $\Delta\epsilon_{339} +0.51$ ,  $\Delta\epsilon_{240} -1.96$ ,  $\Delta\epsilon_{212} +2.57$ .

**Methyl (14S)-14,15-dihydroxylabda-8(17),13(16)-dien-19-oate (15).** Colourless oil.  $C_{21}H_{34}O_4$ ;  $[\alpha]_D +42.9^\circ$  (c 1.56); IR  $\nu_{max}$   $cm^{-1}$ : 3400, 3070, 1725, 1640, 1230, 1190, 1160, 990, 890;  $^1H$  NMR ( $CCl_4$ ):  $\delta$  0.51 (3H, s, Me-10), 1.20 (3H, s, Me-4), 3.61 (3H, s, OMe), 3.65 (2H, m,  $-CH_2-OH$ ), 4.14 (1H, m,  $-CHOH-$ ), 4.50 (1H, br.s H-17), 4.86 (1H, br.s, H-17), 4.95 (1H, br.s, H-16), 5.12 (1H, br.s, H-16). CD ( $CCl_4$ , concn: 1.5 mM **16** and 0.2 mM  $Pr(fod)_3$ ):  $\Delta\epsilon_{317} -1.2$ . **Diacetate:** oil,  $[\alpha]_D +39.3^\circ$  (c 1.21); IR  $\nu_{max}$   $cm^{-1}$ : 3060, 1740, 1720, 1640, 1230, 1150, 1035, 885;  $^1H$  NMR ( $CCl_4$ ):  $\delta$  0.49 (3H, s, Me-10), 1.15 (3H, s, Me-4), 1.97 (3H, s, OAc), 2.02 (3H, s, OAc), 3.55 (3H, s, OMe), 4.07 (2H, m, AB part of an ABX system H-15), 4.48 (1H, br.s H-17), 4.84 (1H, br.s H-17), 4.89 (1H, br.s H-16), 5.02 (1H, br.s H-16), 5.24 (1H, q, X part of the ABX system H-14). MS  $m/e$  (rel. int.): 374 ( $M^+$  -60, 1.5), 314 (13), 299 (5), 255 (16), 254 (14), 239 (13), 121 (100).

**Photooxidation of methyl isocupressate.** To **5** Me ester (220 mg) in *iso*-PrOH (31 ml) 6 mg of Rose Bengal were added and the stirred mixture was exposed to sunlight for 6 hr. The *iso*-PrOH was evapd, MeOH and  $NaBH_4$  (307 mg) were added, the mixture stirred at room temp. for 30 min, worked up as usual and after PLC on Si gel ( $C_6H_6-Et_2O$ , 6:4) gave 46 mg of methyl-14,15-dihydroxylabda-8(17),13(16)-dien-19-oate,  $[\alpha]_D +52.5^\circ$  (c 1.6) with identical  $R_f$ , IR and NMR to **15**.

**Methyl-7 $\alpha$ -hydroxysandaracopimarate (17).** Viscous oil.  $C_{21}H_{32}O_3$ ;  $[\alpha]_D -41.0^\circ$  (c 0.84); IR  $\nu_{max}$   $cm^{-1}$ : 3400, 3070, 1720, 1640, 1245, 995, 905;  $^1H$  NMR:  $\delta$  0.84 (3H, s, Me-10), 1.08 (3H, s, Me-13), 1.22 (3H, s, Me-4), 3.70 (3H, s, OMe), 4.16 (1H, m,  $W_{1/2} = 9$  Hz), 4.80–5.10 (2H, AB part of an ABX system, H-16), 5.55 (1H, br.s, H-14) and 5.60–6.05 (1H, X part of the ABX system, H-15); MS  $m/e$  (rel. int.): 332 ( $M^+$ , 2), 330 (3), 315 (3), 271 (4), 255 (7), 203 (11), 177 (22), 121 (67), 55 (100).

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